AMENDMENTS TO THE CLAIMS:

10/7 26 23 -

The listing of claims shown below will replace all prior versions, and listings, of claims in the Application:

(Currently Amended)	A method of forming MgB ₂ films in-situ on a
substrate comprising the steps:	

- (a) depositing boron onto a surface of the substrate in a <u>depressurized</u> deposition zone;
- (b) moving the substrate into a reaction zone containing pressurized gaseous magnesium, the reaction zone being substantially sealed from the depressurized deposition zone;
 - (c) moving the substrate back into the deposition zone; and
 - (d) repeating steps (a)-(c).

(C) is produced by rotating the substrate on a platen.

(Original) The method of claim 2, wherein the platen is rotated at a rate within the range of about 100 rpm to about 500 rpm.

(Original) The method of claim 1, wherein the substrate is heated to a temperature within the range of about 300°C to about 700°C.

(Original) The method according to claim 1, wherein the substrate is selected from the group consisting of LSAT, LaAlO₃, MgO, SrTiO₃, r-plane sapphire, c-plane sapphire, m-plane sapphire, yttria-stabilized zirconia (YSZ), silicon carbide, polycrystalline alumina, silicon, and stainless steel.

(Currently Amended) The method of claim/1, wherein the reaction zone contains gaseous magnesium at a partial pressure of about 10 mTorr. A MgB₂ film produced by the method of claim 1.

(Original) The method according to claim (1, wherein the reaction zone is coupled to a heated source of magnesium.

(Original) The method according to claim 1, wherein the substrate is a

(Original) The method according to claim 1, wherein the substrate is a tape.

(Original) The method according to claim 1, wherein the method is used to form MgB₂ on a plurality of substrates.

(Currently Amended) The method of claim 1, wherein the boron is evapprated the film of MgB₂ is generated under at a pressure of less than 10⁻⁶ Torr in the

deposition zone.

	\ .	14	(Original)	The method of	claim 1 v	vherein the MgB₂ fi	lm is formed on a
	single	e side d	f the substrat	te.			
<u> </u>							
1	M		Currently A	mended) <u>A meth</u>	od of form	ning MgB ₂ films <i>in-</i> s	situ on a substrate
(/	<u>comr</u>	rising t	he steps:				
	_	(a)	depositing b	oron onto a surfa	ace of the	substrate in a depo	osition zone;
		(b)	moving the	substrate into a r	eaction zo	one containing pres	surized gaseous
	magr	nesium;				·	
		(c)	moving the	substrate back in	to the der	osition zone; and	
		(d)	repeating st	eps (a)-(c);			
	The r	nethod	of claim 1, w	herein the MgB ₂	film is forn	ned on two sides o	/ f the substrate.
		14	(Currently A	mended) A meth	od of form	ning a film of MgB ₂	in-situ comprising
	the st	eps of:					
		provid	ling a rotatab	le platen, the pla	ten being	rotatable within a h	ousing having a
	press	urized I	eaction zone	and a separate	depressur	ized deposition zor	ne, the pressurized
	<u>reacti</u>	on zon	e being subst	antially sealed from	m the de	pressurized depos	tion zone;
		provid	ing an evapo	ration cell operat	ively coup	oled to the pressurfi	<u>/</u> <u>zed</u> reaction zone,
	the ev	/aporat	ion cell conta	ining magnesium);		
		provid	ing a source	of boron dispose	d adjacen	t to the depressuriz	zed deposition zone;
		provid	ing an electro	on beam gun aim	ed at the	source of boron;	
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loading a substrate onto the platen;

rotating the platen;

heating the local environment around the substrate;

heating the evaporation cell so as to produce <u>pressurized</u> gaseous magnesium in the reaction zone; and

evaporating the boron with the electron beam gun.

(Original) The method according to claim 14, wherein the local environment around the substrate is heated to a temperature within the range of about 300°C to about 700°C.

(Original) The method according to claim 14, wherein the evaporation cell is heated to a temperature of at least 550°C.

(Original) The method according to claim 14, wherein the platen is rotated at a rate within the range of about 100 rpm to about 500 rpm.

The method according to claim 14 wherein the substrate is selected from the group consisting of LSAT, LaAlO₃, MgO, SrTiO₃, r-plane sapphire, c-plane sapphire, m-plane sapphire, yttria-stabilized zirconia (YSZ), silicon carbide, polycrystalline alumina, silicon, and stainless steel.

(Original) The method of claim 14, wherein the substrate is a wafer.

∀ ≥∮ .	(Original)	The method	of claim 1	4, wherein t	he substrate	is a tape.	
Y 21	(Original)	The method	of claim	4 wherein t	he step of lo	ading the	plater
comprises lo	pading the pla	ten with a plu	rality of sul	ostrates.			•
27.	(Currently A	mended)	The meth	nod of claim	14, wherein	the boron	ı i <u>s</u>
evaporated :	the film of Mg	B ₂ -is-generate	ed under at	a pressure	of less than	10 ⁻⁶ Torr i	in the
deposition z	one.						
× 33.	(Original)	The method	of claim 14	wherein a	a film of MgB	2 is formed	d on a
single side o	of the substrat	e.					
Comprising		mended)	A method	l of forming	a film of Mg	B ₂ in-situ	
		Ly L				·	
	<u>ling a rotatabl</u>	,		<u>g rotatable</u>	<u>within a hou</u>	sing havin	<u>g a</u>
reaction zon	e and a separ	ate deposition	n zone;				
provid	ling an evapo	ration cell ope	eratively co	upled to the	reaction zo	ne, the	
evaporation	cell containing	<u>ı magnesium;</u>					
provid	ling a source	of boron dispo	osed adjac	ent to thé de	eposition zor	<u>1e;</u>	
•	ling an electro		,				
	g a substrate						
	o the platen						

heating the evaporation cell so as to produce gaseous magnesium in the reaction
zone; and
evaporating the boron with the electron beam gun;
The method of claim 14, further comprising the steps of removing the substrate from the
platen;
turning the substrate over;
loading the substrate onto the platen;
rotating the platen;
heating the local environment around the substrate;
heating the evaporation cell so as to produce pressurized gaseous magnesium in
the reaction zone; and
evaporating the boron with the electron beam gun.
(Currently Amended) The method of claim 14, wherein the reaction zone
contains gaseous magnesium at a partial pressure of about 10 m orr. A MgB ₂ film produced
by the method of claim 14.
(Currently Amended) A method of forming a superconducting film of a known superconducting compound <i>in-situ</i> on a substrate comprising the steps:
(a) depositing one or more elements of the superconductor onto a surface of the
substrate in a depressurized deposition zone having a pressure less than about 10 Torr,
(b) heating a non-gaseous element of the superconductor so as to produce a
8 (7)

heating the local environment around the substrate;

pressurized gaseous phase of the element inside a reaction zone, the reaction zone being substantially sealed from the depressurized deposition zone and being substantially free of oxygen; moving the substrate into the reaction zone containing the pressurized (c) gaseous element; (d) moving the substrate back into the depressurized deposition zone; and (e) repeating steps (a)-(d). (Currently Amended) The method of claim 26, wherein the superconducting film is a superconductor selected from the group consisting of magnesium diboride, YBCO, BSCCO, TBCCO, and HBCCO. (Currently Amended) A method of forming a film of\a known compoxind in-situ on a substrate comprising the steps: (a) depositing one or more elements of the compound onto a surface of the substrate in a one of a plurality of depréssurized deposition zones; (b) heating a non-gaseous element of the compound so as to produce a pressurized gaseous phase of the element inside a plurality of reaction zones, each reaction zone being substantially sealed from the depressurized deposition zones; (c) moving the substrate into a next the reaction zone containing the pressurized gaseous element; (d) moving the substrate back into the a next depressurized deposition zone; and

(e)

repeating steps (a)-(d).

(Original) The method of claim 28, wherein the compound is a superconductor.

(New) The method of claim 26, wherein step (c) further comprises moving the substrate into another reaction zone containing oxygen.

The method of claim 30, wherein the superconducting film is a superconductor selected from the group consisting of YBCO, BSCCO, TBCCO, and HBCCO.